

1. A cubic box is described by a three-dimensional infinite square well with $0 \leq x, y, z \leq a$ and corresponding single-particle energy eigenstates

$$E = \frac{\hbar^2 \pi^2}{2ma^2}(p^2 + q^2 + r^2)$$

where p, q, r are positive integers. Assume that the box contains N non-interacting distinguishable particles of mass m . Let $G(E)$ be the number of states with energy less than E . For $E \gg \hbar^2 \pi^2 / (2ma^2)$ show that $G(E) \approx cE^{\alpha N}$ for some positive constants c and α . Determine the value of α .

[*Hint:* a microstate can be parameterized by $3N$ positive integers and states with energy less than E are those for which these integers lie inside some sphere in the corresponding $3N$ dimensional space. How does the volume of this sphere depend on N ?]

2. Consider two systems, for which the total number of states with energy less than E is given by the functions $G_1(E)$ and $G_2(E)$ with $G_i(E) = c_i E^{\alpha_i N_i}$ where N_1, N_2 are the number of particles in each system and c_i, α_i are positive constants. Using the approximation $S_i(E) \approx k \log G_i(E)$ consider the probability that, in equilibrium, the first system has energy E_1 (where E is the total energy):

$$p(E_1) = \frac{\Omega_1(E_1)\Omega_2(E - E_1)}{\Omega(E)} = \frac{1}{\Omega(E)} \exp[S_1(E_1)/k + S_2(E - E_1)/k]$$

(i) Determine the value E_1^* which maximizes this probability. (ii) By expanding the exponent to second order around E_1^* show that $p(E_1)$ is negligible for $|E_1 - E_1^*| \gg E/\sqrt{N}$ (where N is the total number of particles so $N_1, N_2 \sim N$).

3. A macroscopic body at room temperature (293K) absorbs a photon of visible light with energy 2eV. By what factor does $\Omega(E)$ for the body increase? What would be the answer if the photon originated from a radio transmitter with frequency 10^8 Hz?
4. A system consists of N spin-1/2 particles with fixed positions. Each particle can be in one of two quantum states: spin “up” or “down”, labelled by $s_z \in \{1/2, -1/2\}$. There is a magnetic field of strength B in the z -direction so the energy of a particle is $-\mu B s_z$ where μ is a constant.

(a)(i) How many states are there with N_\uparrow spin “up” particles and $N_\downarrow = N - N_\uparrow$ spin “down” particles? Express the energy of such a state as a function of N_\uparrow and hence calculate the Boltzmann entropy $S(E)$, expressing your answer in terms of N_\uparrow . When

N , N_{\uparrow} and N_{\downarrow} are all large use Stirling's approximation $n! \approx \sqrt{2\pi n} n^{n+1/2} e^{-n}$ to show that

$$S(E) \approx -k \left[(N - N_{\uparrow}) \log \left(\frac{N - N_{\uparrow}}{N} \right) + N_{\uparrow} \log \left(\frac{N_{\uparrow}}{N} \right) \right]$$

Plot S as a function of E .

(ii) Calculate the temperature T as a function of E and plot $1/T$ against E . What happens in the limits $T \rightarrow 0$ and $T \rightarrow \infty$? Show that the temperature is *negative* for $E > 0$.

[Negative temperature occurs if Ω decreases with E , which is the case here for $E > 0$. For most systems, kinetic energy ensures that Ω always increases with E as in Q1 above, so negative temperatures do not usually occur.]

(b) Now consider this system in the canonical ensemble. Show that the partition function is

$$Z = 2^N \cosh^N \left(\frac{\beta \mu B}{2} \right)$$

Find the average energy E and entropy S . Show that your results for $E(T)$ and $S(T)$ agree with the microcanonical calculations of part (a).

(c) The *magnetisation* of the system, is defined by $M = \mu(N_{\uparrow} - N_{\downarrow})/2$. Compute M as a function of (T, B) . The magnetic susceptibility is defined as $\chi \equiv (\partial M / \partial B)_T$. Show that, at high temperature, the system obeys *Curie's Law*: $\chi \sim 1/T$.

5. Consider a system of N interacting spins. At low temperatures, the interactions ensure that the spins are either all aligned or all anti-aligned with the z axis, even in the absence of an external field. At high temperatures, the interactions become less important and the spins can be either aligned or anti-aligned with the z -axis. The heat capacity takes the form

$$C_V = C_{\max} \left(\frac{2T}{T_0} - 1 \right) \text{ for } \frac{T_0}{2} < T < T_0 \text{ and } C_V = 0 \text{ otherwise}$$

Determine C_{\max} .

6. Compute the partition function of a quantum harmonic oscillator with frequency ω and energy levels

$$E_n = \hbar\omega \left(n + \frac{1}{2} \right) \quad n \in \mathbf{Z}$$

Find the average energy E and entropy S as a function of temperature T .

Einstein constructed a simple model of a solid as N atoms, each of which vibrates with the same frequency ω . Treating these vibrations as a harmonic oscillator, show that at high temperatures, $kT \gg \hbar\omega$, the Einstein model correctly predicts the *Dulong-Petit* law for the heat capacity of a solid,

$$C_V = 3Nk$$

At low temperatures, the heat capacity of many solids is experimentally observed to tend to zero as $C_V \sim T^3$. Was Einstein right about this?

7. (a) A quantum violin string can vibrate at frequencies ω , 2ω , 3ω and so on. Each vibration mode can be treated as an independent harmonic oscillator. Ignore the zero point energy, so that the mode with frequency $p\omega$ has energy $E = n\hbar p\omega$, $n \in \{0, 1, 2, \dots\}$. Write an expression for the average energy of the string at temperature T . Show that at large temperatures the free energy is given by,

$$F = -\frac{\pi^2}{6} \frac{k^2 T^2}{\hbar\omega}$$

(Hint: You may need the value $\zeta(2) = \pi^2/6$)

- (b)* Show that the partition function of the quantum violin string can be written as

$$Z = \sum_N p(N) e^{-\beta E_N}$$

where $E_N = N\hbar\omega$ and $p(N)$ counts the number of partitions of N . It can be shown that this formula also applies to a relativistic string if we use $E_N = \sqrt{N}\hbar\omega$. Show that the relativistic string has a maximum temperature, known as the Hagedorn temperature, $kT_{\max} = \sqrt{6}\hbar\omega/2\pi$.

(Hint: Google the Hardy-Ramanujan formula).

8. Consider the Gibbs entropy for a probability distribution p_i on energy eigenstates $|i\rangle$:

$$S = -k \sum_i p_i \log p_i$$

(a) Show that at fixed *average* energy $\langle E \rangle = \sum_i p_i E_i$, the entropy is maximised by the canonical ensemble. Moreover, show that the Lagrange multiplier imposing the constraint is proportional to β , the inverse temperature. Confirm that maximizing the entropy is equivalent to minimizing the free energy.

(b) Show that at fixed average energy $\langle E \rangle$ and average particle number $\langle N \rangle$, the entropy is maximised by the grand canonical ensemble. What is the interpretation of the Lagrange multiplier in this case?

9. Let Z_N be the canonical partition function for N particles. Show that the grand partition function \mathcal{Z} can be written as

$$\mathcal{Z}(\mu, V, T) = \sum_{N=0}^{\infty} \xi^N Z_N(V, T)$$

where $\xi = e^{\mu\beta}$ is called the *fugacity*. (It will be denoted z in the lecture notes but I wanted to save you from having to write three different types of z). Show that

$$\langle N \rangle = \xi \frac{\partial}{\partial \xi} \log \mathcal{Z} \quad , \quad (\Delta N)^2 = \left(\xi \frac{\partial}{\partial \xi} \right)^2 \log \mathcal{Z} \quad .$$

If $Z_N = Z_1^N/N!$ show that $\mathcal{Z}(\xi, V, T) = e^{\xi Z_1(V, T)}$. For this case, show also that

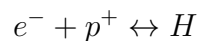
$$\frac{\Delta N}{\langle N \rangle} = \frac{1}{\langle N \rangle^{1/2}} \quad .$$

10. Make use of the fact that the free energy $F(T, V, N)$ of a thermodynamic system must be extensive, to explain why

$$F = V \left(\frac{\partial F}{\partial V} \right)_{T, N} + N \left(\frac{\partial F}{\partial N} \right)_{T, V} \quad .$$

The *Gibbs free energy* is defined as $G = F + pV$. Use the result above for F to show that the Gibbs free energy can be expressed as $G = \mu N$. Explain why this result was to be expected from the scaling behaviour of G .

11. A neutral gas consists of N_e electrons e^- , N_p protons p^+ and N_H hydrogen atoms H . An electron and proton can combine to form Hydrogen,



At fixed temperature and volume, the free energy of the system is $F(T, V; N_e, N_p, N_H)$. We can define a chemical potential for each of the three species as

$$\mu_i = \frac{\partial F}{\partial N_i}$$

By minimizing the free energy, together with suitable constraints on the particle numbers, show that the condition for equilibrium is

$$\mu_e + \mu_p = \mu_H$$

Such reactions usually take place at constant pressure, rather than constant volume. What quantity should you consider instead of F in this case?